

This Page Is Inserted by IFW Operations
and is not a part of the Official Record

BEST AVAILABLE IMAGES

Defective images within this document are accurate representations of the original documents submitted by the applicant.

Defects in the images may include (but are not limited to):

- BLACK BORDERS
- TEXT CUT OFF AT TOP, BOTTOM OR SIDES
- FADED TEXT
- ILLEGIBLE TEXT
- SKEWED/SLANTED IMAGES
- COLORED PHOTOS
- BLACK OR VERY BLACK AND WHITE DARK PHOTOS
- GRAY SCALE DOCUMENTS

IMAGES ARE BEST AVAILABLE COPY.

**As rescanning documents *will not* correct images,
please do not report the images to the
Image Problems Mailbox.**

10/031423

C13 Rec'd PCT/PTO 16 JAN 2002

Substitute Specification, Including Amendment Made

Under PCT Article 34 to Specification and Claims

Title of Invention:

**PROCESS FOR THE QUANTITATIVE DETERMINATION
OF ALKALOIDS SUCH AS COCAINE IN A SOLID
SAMPLE AND REAGENT FOR USE IN SUCH PROCESS**

International Application Number: PCT/IB00/00966

International Filing Date: 14 July 2000 (14.07.2000)

International Publication Number: WO 01/06251 A3

International Publication Date: 25 January 2001 (25.01.2001)

WO 01/06251

13 Rec'd PCT/PTO 16 JAN 2002
PCT/IB00/00966

16/prt/s

PROCESS FOR THE QUANTITATIVE DETERMINATION OF ALKALOIDS SUCH AS COCAINE IN A SOLID SAMPLE AND REAGENT FOR USE IN SUCH PROCESS.

5 This invention relates to a process for the quantitative determination of cocaine and other alkaloids, such as morphine and methadone, in a solid sample, e.g., a sample of hair, by using a screening type approach.

10 The invention also relates to a reagent for use in such process and new diagnostic kits including such reagent among their components.

By "screening type approach" is meant a kind of analysis permitting to analyze in a relatively short time span a relatively large number of samples in a cheap, efficacious and standardized manner. This kind of analysis permits to exclude 15 the negative samples by immediately identifying the samples that do not contain the substance or the entire substance class or those in which said substances are present at a level lower than the threshold or cut-off value.

20 The threshold or cut-off is a practical limit selected to establish if the sample analyzed is positive or negative. The threshold value differs from the limit of detection of the method, that is, the lowest concentration of an analyte that can be determined. In fact, the cut-off value is normally set at a concentration higher than the limit of detection in order to obviate the imprecision of the analysis at values close to the limit of detection.

25 Cut-off values are conventionally established and take into account a multiplicity of factors, such as the capability of using commercially available reagents, the pharmacokinetic properties of the substances and the need to avoid false negatives (see: P.Zuccaro, S.Pichini, I.Altieri and R.Pacifici:
30 Proposta di linee guida per l'analisi delle sostanze d'abus nei liquidi biologici

WO 01/06251

PCT/IB00/00966

in RAPPORTI ISTISAN 96/29 edited by ISTITUTO SUPERIORE DI SANITA').

The substances that the metabolism of the human body accumulates in the hair are numerous. Usually, it is required to detect the presence of alkaloids and other substances of abuse, such as, morphine, methadone and/or cocaine. By the method of this invention further substances can also be detected, such as, those of the following non-limitative list: 6-O- mono-acetyl-morphine, bi-acetyl-morphine, codeine, papaverine, nalorphine, nicotine, cotinine, caffeine, noscarpine, mepivacaine, trimetropin, buprenorphine, pentazocine, methadone metabolyte, benzoylecgonine, amphetamine, methylenedioxymethamphetamine, methylenedioxymethamphetamine, methylenedioxyethylamphetamine, benzodiazepines, barbiturates.

15

Description of the state of the art

Several techniques are known for the determination of the analytes of interest, such as those mentioned above. Specifically, for the analysis of cocaine that is present in hair, gas chromatography (GC) combined with Mass Spectrometry (MS) hereinafter referred to as GC/MS and Radio Immune Assay technique (R.I.A.) are known.

GC or liquid chromatography (LC) combined with MS are methods of resolution, purification or separation and identification of components of complex mixtures of organic or inorganic substances having even strictly similar chemical properties. The separation of substances dissolved in a liquid or fixed on a finely divided or porous solid substance is based on percolation, respectively elution through them of an eluent gas, respectively liquid.

30 When the substances to separate /detect can be made gaseous and as eluent a

WO 01/06251

PCT/IB00/00966

gas is used, GC applies. The latter is a separation method based on the distribution between a solid or liquid stationary phase and a mobile phase made of the gases or vapors to separate, which are carried by a stream of an inert gas.

5

The analytical results are reported in a graph named chromatogram, in which the quantities of the single components present in the mixture and transferred to the eluent gas/liquid are reported versus time. The graph has peaks whose highness is a direct function of the quantity of the specific substance.

10

Chromatographic methods that can be used for confirmatory analyses are GC and LC, the latter being often referred to as High Performance Liquid Chromatography (HPLC). The most commonly employed detectors for GC are electron scattering detectors and phosphor nitrogen detectors.

15

As said above, GC/MS employs a gas chromatograph coupled to a mass spectrograph. By so doing, the separation capability of GC combines with the specificity proper of MS. Therefore, GC/MS represents the method of choice for confirmatory analyses of the above named substances as well as their metabolites (see again "Proposta di linee guida per l'analisi delle sostanze d'abuso nei liquidi biologici" by ISTITUTO SUPERIORE DI SANITA')

20

In the conventional acid hydrolysis, however, cocaine is extracted by hydrochloric acid as it is, i.e. without undergoing transformation into its metabolite benzoylecgonine. Since the amounts of cocaine extracted in this way are very little, it is not possible to determine cocaine by the usual screening methods.

25

Radio Immune Assay (RIA) is based on radio-immunological tests using known amounts of antibodies and of analyt labeled with a radioisotop (generally

WO 01/06251

PCT/IB00/00966

I^{125}). During incubation the labeled analyte and that possibly present in the sample compete for the antibody sites. After precipitation of antigen-antibody complexes and centrifugation the supernatant or the precipitate are transferred to a gamma Geiger counter that measures the radioactivity level.

5 RIA kits are extremely sensitive and allow identification of 1-5 ng of substance per ml. Adoption of automatic instruments for pipetting and counting allows the contemporaneous analysis of numerous samples with response times of 1 to 5 hours. On the other hand, the use of radioactive isotopes requires adequate safety measures. Furthermore, the relatively short

10 half-life of the radioactive isotopes imposes a careful handling of reagents.

RIA does not lend itself to a widespread use for the quick determination of the substances in question in view, first of all, of the high cost of reagents, further increased by liability of the antisera. Secondly, in view of the criticality

15 of the analysis due to the need of operating with radioactive materials that are dangerous for the analysts and also in view of the required times which are relatively long. RIA also requires the availability of rooms properly equipped and shielded, as well as particular care in handling waste materials and in their disposal.

20

Therefore, RIA is scarcely employed in this sort of analyses. The prior art shows therefore the impracticality of determining cocaine present in the hair by the so-called screening type approach, as described above, i.e., by very quick and cheap techniques as are used for the determination of the same

25 substances in urine. This is mainly due to the fact that in the conventional screening-type approach what is analyzed is not cocaine, but its metabolite benzoylecgonine. The fact that this substance does not exist in nature by itself, but only as the metabolite of cocaine and the fact that the reaction of transformation of cocaine into benzoylecgonine, i.e. the transformation of the

30 ester group into an hydroxyl group, is irreversible in an alkaline environment

WO 01/06251

PCT/IB00/00966

render the determination of benzoylecgonine a necessary and sufficient condition for the detection of cocaine.

Benzoylecgonine is found in the hair in a ratio of 1 to 4 or even 1 to 10 and more with respect to cocaine. Therefore, when the amount of cocaine is very low or close to the cut-off values (0.2-0.1 ng cocaine per mg hair) screening cannot be performed because of the low amount of this metabolite (0.05-0.025 ng benzoylecgonine per mg hair), such amount being undetectable by the screening apparatuses available in the laboratories.

10

Summary of the invention

The main object of this invention is to overcome the above mentioned drawbacks, i.e. the impossibility to dose the cocaine present in a solid sample by the conventional screening-type approach.

In accordance with the invention, this object is achieved by a screening-type procedure for the quantitative determination of cocaine and other alkaloids which are present in a solid sample which includes the following steps:

- 20 - providing a solid sample in a finely divided and/or powdered form;
- completely immersing the sample into a liquid reagent providing a concentration of hydroxyl groups in the range of from 0.0001 to 5 M;
- maintaining the sample immersed in the liquid at a temperature of 10 to 250°C for a period of time from a few seconds to 48 hours; and,
- 25 - analyzing the liquid separated from the solid with a conventional kit for the determination of the said substances in urine.

Preferred aspects of the invention include using hair as the solid material forming the solid sample; using a buffer as the reagent which provides 30 hydroxyl groups, more preferably an ammonia buffer, most preferably a buffer

WO 01/06251

PCT/IB00/00966

that is hereinafter referred to as VMA; and heating the sample immersed in the liquid at a temperature of about 100 to 150°C for about one hour.

In accordance with another aspect of the invention a process is provided
5 which includes the additional steps of:

- arranging samples by increasing concentrations of the substances of interest; and
- performing confirmation analyses with known techniques, such as, GC or GC/MS.

10

In one embodiment of the invention, the process may provide the following steps:

- providing a sample made of about 50 to 300 mg of finely divided and/or powdered material;
- 15 - adding in the test tube containing the said sample a suitable liquid reagent until the sample is completely immersed, said reagent being capable of performing extraction and transformation of cocaine into benzoylecgonine and at the same time of extracting other similar substances which are present in the sample
- 20 - if necessary, agitating the test tube to facilitate immersion of the sample;
- heating the contents of the test tube to a temperature T₁ for a time interval t₁ by keeping the test tube immersed in a thermostated bath or by placing it in an oven;
- cooling the test tube to room temperature;
- 25 - taking the liquid and transferring it into a test tube suitable for a screening type instrument;
- performing the screening by using a kit of reagents for the determination of the said substances in urine;
- reading the data resulting from the first level instrumentation to verify the
- 30 concentration values with respect to the cut-off limit; and

WO 01/06251

PCT/IB00/00966

- contemporaneously determining the amount(s) of substance(s) present.

In accordance with another aspect of the invention a reagent is provided for use in the above mentioned process which in the most preferred embodiment

5 has the following formula:

$$\text{VMA} = 0.2 \text{ M } (\text{NH}_4)_2\text{HPO}_4 + 5\text{ml/L } 25\% \text{ NH}_4\text{OH}$$

In the above formula $(\text{NH}_4)_2\text{HPO}_4$ is dibasic ammonium phosphate and NH_4OH 10 is ammonium hydrate. In fact, 5ml/L NH_4OH give a 0.07 M concentration of hydroxyl groups, which is comprised in the range giving 100% conversion.

In addition to the main advantage of the invention as outlined above, another 15 advantage is that the confirmation analyses (e.g., GC/MS) need to be performed only on those samples that have been determined to be positive at the initial screening whereas, when conventional techniques are used, all samples must be analyzed since there is no kit available for determination of cocaine, but only for determination of benzoylecgonine.

20 In average, the analysis of 25 samples requires two hours for the preparation of the samples plus one week for the confirmation analyses of all (both positive and negative) samples.

On the other hand, by applying the present invention, an analyst will require 25 for the same 25 samples two hours for samples preparation plus 30 minutes for the screening-type analysis plus the time necessary for performing the confirmation analyses. However, the latter need to be performed only on those samples which have been determined to be positive.

30 Since, in the average, positive samples are between 0 and 20% and since

WO 01/06251

PCT/IB00/00966

most of the time required is spent in the confirmation analyses, the procedure according to the present invention allows saving of substantial time which is estimated at about 70 to 90% of the time required for the analysis of the 25 samples taken into consideration.

5

Another advantage according to this invention is that the possible dragging of the active substances from one sample to another in the confirmation analyses is eliminated or minimized because it is possible to arrange samples in order of increasing concentration of the substances of interest, therefore 10 proceeding to confirmation analyses starting from those samples having the lowest concentration and then with those having higher and higher concentrations.

In this way, it is completely eliminated the possibility that a sample with high 15 concentration of, e.g. cocaine, leaves a trace of it in the instrument used for the analysis and has an impact on the determination of the same substance in the following sample, perhaps having a concentration just below the cut-off.

Still another advantage of this invention is that it allows to search and 20 determine at the same time and in the same sample not only cocaine but also other substances, such as, morphine by applying the same cut-off limit or else methadone by modifying the limit of cut-off.

One further advantage is that the procedure according to this invention offers 25 a higher guarantee to the analyst because positive outcome is confirmed by two different analytical methods.

In a further aspect of the invention a diagnostic kit is provided including the reagent described above as one of its components.

30

WO 01/06251

PCT/IB00/00966

Still further additional advantages will appear from the reading of the following detailed description and the following non-limitative examples.

Brief Description of the Drawings

5

Figure 1 is a chromatogram obtained from an apparatus performing gas chromatography on the sample of Table 2;

10 Figure 2 is the graph obtained by Mass Spectrometry on the first sample after acid hydrolysis and shows presence of cocaine;

Figure 3 is the Mass Spectrum of the same first sample after treatment according to the present invention and shows presence of benzoylecgonine;

15 Figure 4 is the Mass Spectrum of the same first sample after treatment according to the present invention and shows presence of morphine;

Figure 5 is a chromatogram similar to that of Figure 1 and refers to the first sample treated according to this invention as shown in Table 3;

20

Figure 6 is similar to the MS of Figure 3 showing presence of benzoylecgonine;

Figure 7 is similar to the MS of Figure 3 showing presence of morphine;

25

Figure 8 is similar to Figure 1 but relates to a second sample;

Figure 9 is similar to Figure 5 but relates to a second sample;

30 Figures 10 and 11 are respectively similar to Figures 8 and 9 but relate to a

WO 01/06251

PCT/IB00/00966

third sample;

Figures 12 and 13 are respectively similar to Figures 8 and 9 but relate to a fourth sample;

5

Figure 14 is a diagram showing transformation of cocaine into its metabolite benzoylecgonine;

Figure 15 is a graph showing concentration of cocaine versus concentration of OH⁻ when the reaction of transformation of cocaine into benzoylecgonine occurs at 100° C for 1 hour; and

Figure 16 is similar to Figure 15 and shows the case in which the reaction occurs at 150°C for 1 hour.

15

Detailed description of invention embodiments

The process of this invention substantially is a process for the quantitative extraction and transformation of cocaine into benzoylecgonine and for the extraction of similar alkaloids, in particular toxic substances of abuse and/or drugs, which are present in a sample, prepared starting from a solid material.

In the following detailed example, the sample is obtained starting from finely divided hair.

25

The procedure allows, in particular, the dosage of cocaine by a screening type technique, at the same time allowing the dosage of possible other toxic substances and the like present in the hair, by using kits of reagents available in commerce and intended for the analysis in urine.

30 The following steps are carried out:

WO 01/06251

PCT/IB00/00966

- crush hair in fragments of two-three mm length;
- weigh the fragments to form a sample having a weight of 50 to 300 mg;
- wash the sample with methanol in a closed test tube at room temperature to eliminate possible external substances that might interfere with the results,

5 such as, for example, traces of drugs external to the hair that have deposited on it because of its presence in the air (by so doing, it is possible to distinguish if the hair was that of a handler or that of the consumer of the drug);

- Repeat the step of washing with methanol, if necessary;

10 - wash the sample with ethanol in a closed test tube to eliminate traces of methanol or water;

- dry in an oven at about 45°C under flow of inert gas, such as, nitrogen;
- add into the test tube 0.5 to 2 ml of the reagent as defined above and shake if necessary;

15 - heat the test tube to 100°C and maintain at this temperature for 1 hour, e.g. by means of a thermostatic bath or an oven;

- cool the test tube to room temperature, e.g., by immersing the tube into cold water or simply leaving it at room temperature for a suitable time span;
- take the liquid and pour into a test tube of the kind used for urine

20 examination (if necessary, centrifuge the test tube to eliminate turbidity);

- insert the tube in a "first-level screening apparatus" for the quantitative determination of the substances sought for (benzoyllecgonine, morphine, methadone, etc);
- adjust settings of the first-level apparatus in a way suitable for small

25 amounts (this may be necessary if the apparatus is alternatively used for determination in hair or urine);

- perform screening -type analyses using the reagents provided with the kit normally employed for the examination of urine;
- read data resulting from the first-level analyses and establish positivity or

30 negativity with reference to the cut-off value.

WO 01/06251

PCT/IB00/00966

The above procedure may then be complemented with the following additional steps when confirmation analyses are required:

- arrange the samples in the order of increasing concentration of the sought for substances, determined as above described;
- 5 - perform confirmation analyses by analyzing the samples taken in the order of the arrangement (in this way the above mentioned problems of dragging traces of drugs from one sample to the other are overcome).

10 The invention process provides for the use of a reagent (hereinafter referred to as VMA) which is a buffer solution. The buffer serves for transforming the cocaine present in the hair into its metabolite benzoylecgonine, as shown in the scheme of Figure 14. Cocaine, in the presence of hydroxyl groups and at a suitable temperature first is transformed into an intermediate product and then into benzoylecgonine plus methanol.

15

20 Buffer solutions are obtained by reacting a salt with its weak base. These solutions have a stable pH; therefore the VMA reagent is able to produce hydroxyl groups in a steady way. The use of solutions in which the production of hydroxyl groups is not regular creates problems when the cocaine concentration is close to the cut-off limit.

As said above, the composition of the buffer for use in the process is preferably the following:

25

$$\text{VMA} = 0.2 \text{ M } (\text{NH}_4)_2\text{HPO}_4 + 5\text{ml/L } 25\% \text{ NH}_4\text{OH}$$

30 Alternatively, VMA may be replaced by solution in which the component being the source of hydroxyl groups is selected among the following non-limitative list of substances: aluminum hydr xid , barium hydroxide octahydrate, benzyltriethylammonium hydroxide, benzyltrimethylammonium

WO 01/06251

PCT/IB00/00966

hydroxide, calcium hydroxid , phenylhydrargirium hydroxide, lithium hydroxide, lithium hydroxide monohydrate, magnesium hydroxide, p potassium hydroxide, potassium hydroxyantimoniate, sodium hydroxide, sodium hydroxide monohydrate, strontium hydroxyde octahydrate,
5 tetramethylammonium hydroxide, tetrapropylammonium hydroxide, trimethylvinylammonium hydroxide.

As solvent, any of the following can be used in alternative:

ethanol, methanol, water, monobasic ammonium phosphate, ammonium acetate, ammonium benzoate, ammonium bicarbonate, ammonium bichromate, ammonium bisulphate, ammonium bromide, ammonium carbamate, ammonium carbonate, ammonium citrate bibasic, ammonium chromate, ammonium iodide, molibdate, ammonium monovanadate, ammonium nitrate, ammonium oxalate monohydrate, ammonium persulphate,
10 ammonium sulphate, ammonium sulphamate, ammonium sulphite, ammonium sulphide, ammonium tartrate, ammonium thiocyanate, ammonium thioglycolate, ammonium thiosulphate, ammonium chloride, sodium phosphate monobasic, sodium phosphate bibasic, potassium phosphate monobasic, potassium phosphate bibasic.
15

20

Figure 15 shows a graph in which the percentage of cocaine transformed into benzoylecgonine is reported versus the concentration of OH⁻ when the reaction is carried out at the temperature of 100°C for 1 hour. From the graph it appears evident that the transformation is maintained at high levels (at least
25 70%) for a range of OH⁻ concentrations of from 0.03 to 0.5 M.

From the graph of Figure 16, which shows hydroxyl concentrations after, respectively, 0, 15, 30 and 60 minutes in a reaction carried out at 150°C; it appears evident that the percentage of transformed cocaine is high only
30 around the abscissa point of 15 minut s.

WO 01/06251

PCT/IB00/00966

From all of the above and numerous other experiments the optimal values for the reaction temperature and time result to be, respectively, 100°C and 1 hour.

- 5 The examples reported below show the quantitative determination of cocaine in hair performed both with known techniques and with the process of the invention. Analyses have been carried out using the following instruments: for the screening analyses, the ROCHE instrumentation named "COBAS MIRA PLUS" which uses reagents provided by the same company and named
- 10 "ABUSCREEN ON LINE"; for GC/MS the instrument provided by the company VARIAN which is named "SATURN GC/MS", model 4D.

EXAMPLES

- 15 Before proceeding to the various screening analyses the apparatus has been controlled with a sample positive to both cocaine and morphine in order to verify feasibility of the methodology.

- 20 As can be seen from Table 1 the sample resulted positive to both cocaine and morphine:

TABLE 1

SUBSTANCE	CONCENTRATION	POSITIVITY/NEGATIVITY
MORPHINE	0.16 ng/ml	POSITIVE
COCAINE	0.15 ng/ml	POSITIVE

- 25 The concentration values detected by the apparatus resulted to be 0.16 for morphine and 0.15 for cocaine, in line with the expected values for both substances present in the control sample.

WO 01/06251

PCT/IB00/00966

EXAMPLE 1

In this example a sample has been analyzed which resulted, in the end, to be
5 positive to both cocaine and morphine.

The sample was subjected to conventional analysis with acid hydrolysis and
then subjected to screening analysis which gave the following results:

10

TABLE 2

15

SUBSTANCE	CONCENTRATION	POSITIVITY/NEGATIVITY
MORPHINE	0.13 ng/ml	POSITIVE
COCAINE	0.01 ng/ml	NEGATIVE

As can be seen, acid hydrolysis was able to extract an amount of morphine
higher than the limit, but the amount of cocaine resulted to be too little to
allow detection by the screening apparatus. The reason is that cocaine was
20 extracted as such and not transformed into its metabolite benzoylecgonine,
which latter is just what present screening methods detect.

The same sample was then subjected to confirmatory analysis by GC/MS with
the results reported in Figures 1 to 4. Specifically, from Figure 1 it can be
25 seen that the chromatogram of the first sample shows a high peak at the
position of cocaine and a very small peak at the position of benzoylecgonine.
The same Figure 1 also shows confirmation of the presence of morphine.

Additional confirmations of the presence of both cocaine and morphine result
30 from the graphs of Figures 2, 3 and 4.

WO 01/06251

PCT/IB00/00966

The sample of Example 1 was then subjected to the procedure of this invention to give the results shown in Table 3.

5

TABLE 3

SUBSTANCE	CONCENTRATION	POSITIVITY/NEGATIVITY
MORPHINE	0.23 ng/ml	POSITIVE
COCAINE	0.37 ng/ml	POSITIVE

It is immediately clear that the values detected for both morphine and cocaine are higher than the respective limits, therefore the sample is positive for both.

15 The high value detected for cocaine is to be ascribed to the benzoylecgonine that has been produced during the transformation reaction.

20 The same sample of Example 1 treated with the process of this invention has been subjected to GC/MS confirmation analyses with the results reported in Figures 5 to 7. As can be seen from Figure 5, the cocaine peak practically cannot be seen anymore because cocaine has been completely transformed in benzoylecgonine, the peak of which is well apparent in the same Figure 5.

EXAMPLE 2

25

Again in this example a sample which, in the end, resulted to be positive to both cocaine and morphine has been analyzed first by the conventional analytical method of acid hydrolysis and screening with the results reported in Table 4.

30

WO 01/06251

PCT/IB00/00966

TABLE 4

<u>SUBSTANCE</u>	<u>CONCENTRATION</u>	<u>POSITIVITY/NEGATIVITY</u>
MORPHINE	0.32 ng/ml	POSITIVE
COCAINE	0.03 ng/ml	NEGATIVE

Like in the preceding Example, acid hydrolysis was able to extract an amount of morphine higher than the limit, but the quantity of cocaine extracted was

10 too little to allow detection by the apparatus.

The sample was then subjected to GC/MS confirmatory analyses to give the results reported in Figure 8 where it can be seen that the chromatogram shows a high peak at the position of cocaine and a negligible peak at the 15 position of benzoylecgonine.

The sample of Example 2 was then subjected to the invention process and to screening analyses with the results reported in Table 5.

TABLE 5

<u>SUBSTANCE</u>	<u>CONCENTRATION</u>	<u>POSITIVITY/NEGATIVITY</u>
MORPHINE	0.28 ng/ml	POSITIVE
COCAINE	0.99 ng/ml	POSITIVE

Both the cocaine and the morphine values are higher than the respective limits, therefore the sample is positive to both.

WO 01/06251

PCT/IB00/00966

The sample of this Example, treated according to the invention process, was also subjected to GC/MS confirmation analyses with the results reported in Figure 9 where it can be seen that the cocaine peak is not practically present anymore whereas the benzoylecgonine peak is high.

5

EXAMPLE 3

In this Example a sample has been analyzed which, at the end, resulted to be positive to cocaine only and not to morphine.

10

This sample was first subjected to acid hydrolysis and screening by the conventional methods to give the results reported in Table 6.

15

TABLE 6

20

<u>SUBSTANCE</u>	<u>CONCENTRATION</u>	<u>POSITIVITY/NEGATIVITY</u>
MORPHINE	0.02 ng/ml	NEGATIVE
COCAINE	0.01 ng/ml	NEGATIVE

25

30

By acid hydrolysis a quantity of morphine lower than the limit was extracted. The quantity of cocaine was also too little to allow the equipment to detect it. By using the methods known in the art, therefore, nothing can be said as to the positivity or negativity of the sample and it was necessary to make recourse to the GC/MS confirmation analyses that gave the results reported in Figure 10. In such Figure, the chromatogram of the sample of this Example shows a high peak at the position of cocaine and a negligible peak at the position of benzoylecgonine. This confirms once more that acid hydrolysis extracted cocaine as such without any transformation.

WO 01/06251

PCT/IB00/00966

The same sample of this Example has also been treated according to the invention process and subjected to screening analysis. Results are reported in Table 7.

5

TABLE 7

10

<u>SUBSTANCE</u>	<u>CONCENTRATION</u>	<u>POSITIVITY/NEGATIVITY</u>
MORPHINE	0.03 ng/ml	NEGATIVE
COCAINE	0.13 ng/ml	POSITIVE

15

It is immediately evident that while the value of morphine is lower than the limit, that of cocaine is slightly higher than the cut-off and therefore the sample is only positive to cocaine.

20

This same sample, after treatment with the invention process, has been subjected to GC/MS confirmation analyses with the results shown in Figure 11. In this Figure, the peak of cocaine has practically disappeared because the same was completely transformed into benzoylecgonine, the peak of which is, instead, very high and well apparent.

EXAMPLE 4

25

This fourth Example also analyzes a sample which, at the end, resulted positive to cocaine and not to morphine.

Analyses were carried out by applying the same scheme as in the preceding Examples.

30

WO 01/06251

PCT/IB00/00966

Table 8 shows that upon analysis with acid hydrolysis the sample seems to be completely negative to both morphine and cocaine, whereas Figure 12 indicates the presence of cocaine in an amount higher than the limit.

TABLE 8

S	<u>SUBSTANCE</u>	<u>CONCENTRATION</u>	<u>POSITIVITY/NEGATIVITY</u>
MORPHINE	0.02 ng/ml	NEGATIVE	
COCAINE	0.05 ng/ml	NEGATIVE	

The same sample, treated with the invention process (VMA reagent) already on the first screening analysis shows a presence of cocaine higher than the limit, see Table 9.

15 TABLE 9

15	<u>SUBSTANCE</u>	<u>CONCENTRATION</u>	<u>POSITIVITY/NEGATIVITY</u>
MORPHINE	0.02 ng/ml	NEGATIVE	
COCAINE	0.64 ng/ml	POSITIVE	

As usual, this result was confirmed by GC/MS analysis as shown in Figure 13. The big peak of benzoylecgonine indicates that, before the treatment, the sample contained cocaine in a percentage higher than the cut-off.

The instant application is based upon Italian patent application VR99A000059, filed on 20 July 1999, the disclosure of which is hereby expressly incorporated by reference thereto, and the priority of which is hereby claimed.